

CRASH COURSE CHEMISTRY PERIODIC TABLE

CRASH COURSE CHEMISTRY PERIODIC TABLE SERVES AS AN ESSENTIAL FOUNDATION FOR UNDERSTANDING THE ELEMENTS THAT COMPOSE THE UNIVERSE AND THEIR INTERACTIONS. THIS ARTICLE PROVIDES A COMPREHENSIVE OVERVIEW OF THE PERIODIC TABLE, FOCUSING ON ITS STRUCTURE, HISTORY, AND SIGNIFICANCE WITHIN THE REALM OF CHEMISTRY. BY EXPLORING THE ORGANIZATION OF ELEMENTS, PERIODIC TRENDS, AND KEY GROUPS, READERS GAIN INSIGHT INTO THE PRACTICAL APPLICATIONS AND THEORETICAL PRINCIPLES THAT GOVERN CHEMICAL BEHAVIOR. THIS CRASH COURSE CHEMISTRY PERIODIC TABLE GUIDE IS TAILORED FOR LEARNERS SEEKING A THOROUGH YET ACCESSIBLE INTRODUCTION TO ONE OF CHEMISTRY'S MOST VITAL TOOLS. THE FOLLOWING SECTIONS WILL COVER THE ORIGINS OF THE PERIODIC TABLE, ITS LAYOUT AND CLASSIFICATIONS, PERIODIC TRENDS, AND IMPORTANT ELEMENT GROUPS TO FACILITATE A WELL-ROUNDED UNDERSTANDING.

- HISTORY AND DEVELOPMENT OF THE PERIODIC TABLE
- STRUCTURE AND ORGANIZATION OF THE PERIODIC TABLE
- PERIODIC TRENDS AND THEIR SIGNIFICANCE
- KEY GROUPS AND THEIR CHARACTERISTICS
- APPLICATIONS OF THE PERIODIC TABLE IN CHEMISTRY

HISTORY AND DEVELOPMENT OF THE PERIODIC TABLE

EARLY ATTEMPTS AT ELEMENT CLASSIFICATION

THE JOURNEY TO THE MODERN PERIODIC TABLE BEGAN IN THE EARLY 19TH CENTURY WITH SCIENTISTS ATTEMPTING TO CATEGORIZE KNOWN ELEMENTS BASED ON THEIR PROPERTIES. EARLY CHEMISTS LIKE JOHANN D[?]BEREINER IDENTIFIED GROUPS OF ELEMENTS WITH SIMILAR CHARACTERISTICS, KNOWN AS TRIADS. HOWEVER, A COMPREHENSIVE AND SYSTEMATIC ARRANGEMENT WAS STILL LACKING, PROMPTING FURTHER INVESTIGATION.

MENDELEEV'S PERIODIC TABLE

DMITRI MENDELEEV IS CREDITED WITH CREATING THE FIRST WIDELY RECOGNIZED PERIODIC TABLE IN 1869. HE ARRANGED ELEMENTS IN ORDER OF INCREASING ATOMIC MASS AND GROUPED THEM BY SIMILAR CHEMICAL PROPERTIES. MENDELEEV'S TABLE WAS REVOLUTIONARY BECAUSE IT PREDICTED THE EXISTENCE AND PROPERTIES OF UNDISCOVERED ELEMENTS, LEAVING GAPS FOR FUTURE DISCOVERIES WHICH WERE LATER CONFIRMED.

MODERN PERIODIC TABLE AND ATOMIC NUMBER

THE MODERN PERIODIC TABLE IS ORGANIZED BY ATOMIC NUMBER RATHER THAN ATOMIC MASS, A SHIFT INITIATED BY HENRY MOSELEY'S WORK IN THE EARLY 20TH CENTURY. THIS CHANGE RESOLVED INCONSISTENCIES IN MENDELEEV'S ARRANGEMENT AND ESTABLISHED THE PERIODIC LAW: THE PROPERTIES OF ELEMENTS RECUR PERIODICALLY WHEN ARRANGED BY INCREASING ATOMIC NUMBER.

STRUCTURE AND ORGANIZATION OF THE PERIODIC TABLE

PERIODS AND GROUPS

THE PERIODIC TABLE IS STRUCTURED INTO ROWS CALLED PERIODS AND COLUMNS CALLED GROUPS OR FAMILIES. EACH PERIOD CORRESPONDS TO THE FILLING OF A DIFFERENT ELECTRON SHELL, WHILE GROUPS CONTAIN ELEMENTS WITH SIMILAR VALENCE ELECTRON CONFIGURATIONS, RESULTING IN COMPARABLE CHEMICAL PROPERTIES.

BLOCKS OF THE PERIODIC TABLE

THE TABLE IS DIVIDED INTO FOUR BLOCKS BASED ON ELECTRON SUBSHELL FILLING:

- **S-BLOCK:** GROUPS 1 AND 2, INCLUDING HYDROGEN AND HELIUM, CHARACTERIZED BY THE FILLING OF S ORBITALS.
- **P-BLOCK:** GROUPS 13 TO 18, CONTAINING ELEMENTS FILLING P ORBITALS.
- **D-BLOCK:** TRANSITION METALS IN GROUPS 3 TO 12, WHERE D ORBITALS ARE BEING FILLED.
- **F-BLOCK:** LANTHANIDES AND ACTINIDES, FILLING F ORBITALS TYPICALLY PLACED BELOW THE MAIN TABLE.

ELEMENT CATEGORIES

ELEMENTS ARE BROADLY CLASSIFIED INTO METALS, NONMETALS, AND METALLOIDS BASED ON THEIR PHYSICAL AND CHEMICAL PROPERTIES. METALS, FOUND MOSTLY ON THE LEFT AND CENTER OF THE TABLE, ARE CONDUCTIVE AND MALLEABLE. NONMETALS, LOCATED ON THE RIGHT SIDE, TEND TO BE INSULATORS AND ARE MORE CHEMICALLY DIVERSE. METALLOIDS EXHIBIT INTERMEDIATE PROPERTIES AND ARE SITUATED ALONG THE STAIR-STEP LINE DIVIDING METALS AND NONMETALS.

PERIODIC TRENDS AND THEIR SIGNIFICANCE

ATOMIC RADIUS

ATOMIC RADIUS REFERS TO THE SIZE OF AN ATOM AND VARIES IN A PREDICTABLE MANNER ACROSS THE PERIODIC TABLE. GENERALLY, ATOMIC RADIUS DECREASES ACROSS A PERIOD FROM LEFT TO RIGHT DUE TO INCREASING NUCLEAR CHARGE PULLING ELECTRONS CLOSER. CONVERSELY, ATOMIC RADIUS INCREASES DOWN A GROUP AS ADDITIONAL ELECTRON SHELLS ARE ADDED.

IONIZATION ENERGY

IONIZATION ENERGY IS THE ENERGY REQUIRED TO REMOVE AN ELECTRON FROM AN ATOM IN ITS GASEOUS STATE. IT INCREASES ACROSS A PERIOD BECAUSE THE NUCLEAR CHARGE STRENGTHENS, MAKING ELECTRONS HARDER TO REMOVE. IONIZATION ENERGY DECREASES DOWN A GROUP AS ELECTRONS ARE FURTHER FROM THE NUCLEUS AND EXPERIENCE MORE SHIELDING.

ELECTRONEGATIVITY

ELECTRONEGATIVITY MEASURES THE TENDENCY OF AN ATOM TO ATTRACT ELECTRONS IN A CHEMICAL BOND. IT GENERALLY INCREASES ACROSS A PERIOD AND DECREASES DOWN A GROUP. ELEMENTS WITH HIGH ELECTRONEGATIVITY, SUCH AS FLUORINE,

STRONGLY ATTRACT BONDING ELECTRONS, INFLUENCING MOLECULE POLARITY AND REACTIVITY.

ELECTRON AFFINITY

ELECTRON AFFINITY DESCRIBES THE ENERGY CHANGE WHEN AN ATOM GAINS AN ELECTRON. ELEMENTS TOWARD THE RIGHT OF THE PERIODIC TABLE TYPICALLY HAVE HIGHER ELECTRON AFFINITIES, MAKING THEM MORE LIKELY TO ACCEPT ELECTRONS DURING CHEMICAL REACTIONS, WHICH IS A KEY FACTOR IN FORMING ANIONS.

KEY GROUPS AND THEIR CHARACTERISTICS

ALKALI METALS (GROUP 1)

ALKALI METALS INCLUDE LITHIUM, SODIUM, POTASSIUM, AND OTHERS. THEY ARE HIGHLY REACTIVE METALS WITH A SINGLE VALENCE ELECTRON, MAKING THEM EAGER TO LOSE THAT ELECTRON AND FORM $+1$ IONS. THESE ELEMENTS ARE SOFT, HAVE LOW MELTING POINTS, AND REACT VIGOROUSLY WITH WATER.

ALKALINE EARTH METALS (GROUP 2)

GROUP 2 ELEMENTS SUCH AS MAGNESIUM AND CALCIUM HAVE TWO VALENCE ELECTRONS AND FORM $+2$ IONS. THEY ARE LESS REACTIVE THAN ALKALI METALS BUT STILL READILY ENGAGE IN CHEMICAL REACTIONS, PARTICULARLY WITH OXYGEN AND WATER UNDER CERTAIN CONDITIONS.

HALOGENS (GROUP 17)

HALOGENS LIKE FLUORINE, CHLORINE, AND BROMINE ARE HIGHLY REACTIVE NONMETALS WITH SEVEN VALENCE ELECTRONS. THEY TEND TO GAIN ONE ELECTRON TO ACHIEVE A STABLE OCTET, FORMING -1 IONS. HALOGENS EXIST IN VARIOUS STATES AT ROOM TEMPERATURE AND ARE COMMONLY FOUND IN SALTS.

NOBLE GASES (GROUP 18)

NOBLE GASES, INCLUDING HELIUM, NEON, AND ARGON, ARE CHARACTERIZED BY THEIR FULL VALENCE ELECTRON SHELLS, RENDERING THEM CHEMICALLY INERT UNDER MOST CONDITIONS. THEIR STABILITY MAKES THEM USEFUL IN LIGHTING, WELDING, AND AS INERT ATMOSPHERES IN CHEMICAL REACTIONS.

TRANSITION METALS

TRANSITION METALS OCCUPY THE D-BLOCK AND ARE KNOWN FOR THEIR VARIABLE OXIDATION STATES AND ABILITY TO FORM COMPLEX IONS. THESE METALS, SUCH AS IRON, COPPER, AND NICKEL, ARE WIDELY USED IN INDUSTRY AND CATALYSIS DUE TO THEIR UNIQUE ELECTRONIC CONFIGURATIONS AND BONDING PROPERTIES.

APPLICATIONS OF THE PERIODIC TABLE IN CHEMISTRY

Predicting Chemical Behavior

The periodic table is an invaluable tool for predicting how elements will behave in chemical reactions. By understanding an element's position, chemists can infer its reactivity, common oxidation states, and the types of compounds it forms, facilitating the design of experiments and synthesis.

Identifying Element Properties

The arrangement of the periodic table allows for quick identification of physical and chemical properties such as metallic character, electronegativity, and atomic radius. This knowledge is crucial in materials science, pharmacology, and environmental chemistry.

Facilitating Education and Research

As a foundational learning resource, the periodic table supports chemistry education at all levels. In research, it guides the discovery of new elements and the development of novel compounds by providing a framework for understanding element relationships and trends.

Industrial and Practical Uses

The periodic table aids industries in selecting appropriate elements for manufacturing, energy production, and technology development. For example, transition metals are essential in catalysis, while noble gases are used in lighting and electronics.

Frequently Asked Questions

What is the Crash Course Chemistry Series?

Crash Course Chemistry is an educational video series that explains key concepts in chemistry in an engaging and accessible way, often used by students and educators.

How does Crash Course explain the periodic table?

Crash Course explains the periodic table by discussing the arrangement of elements based on atomic number, periodic trends, and the organization into groups and periods that reflect element properties.

What are the main groups of elements highlighted in Crash Course Chemistry's periodic table episodes?

The main groups include alkali metals, alkaline earth metals, transition metals, halogens, noble gases, lanthanides, and actinides, each with distinct chemical properties.

Why is the periodic table important in chemistry according to Crash Course?

The periodic table is important because it organizes elements in a way that reveals patterns in their properties and behaviors, helping predict reactions and understand chemical behavior.

How does Crash Course describe periodic trends?

Crash Course describes periodic trends such as atomic radius, ionization energy, and electronegativity, explaining how these properties change across periods and down groups.

What role do electrons play in the periodic table as explained by Crash Course?

Electrons, particularly valence electrons, determine an element's chemical properties and reactivity, which is why elements are grouped based on similar electron configurations.

Does Crash Course Chemistry cover the history of the periodic table?

Yes, Crash Course Chemistry includes the history of the periodic table, highlighting contributions from scientists like Dmitri Mendeleev who first organized elements based on atomic mass and properties.

How can watching Crash Course Chemistry help students understand the periodic table?

Watching Crash Course Chemistry can help students by breaking down complex concepts into simple explanations, using visuals and examples to improve comprehension of the periodic table and its significance.

Are there any experiments or demonstrations related to the periodic table in Crash Course Chemistry?

While Crash Course Chemistry primarily uses animations and explanations, it often references classic experiments and demonstrations that illustrate element properties and periodic trends.

Additional Resources

1. *The Periodic Table: A Visual Guide to the Elements*

This book offers a beautifully illustrated exploration of the periodic table, presenting each element with detailed images and fascinating facts. It bridges the gap between science and art, making the chemistry of elements accessible and engaging for readers of all ages. The visual approach helps reinforce the relationships and trends within the periodic table.

2. *Elements: The Periodic Table in 50 Elements*

Authored by a leading chemist, this book breaks down the complex world of chemistry into 50 essential elements that have shaped our understanding of matter. Each chapter delves into the history, properties, and applications of an element, providing context for its place in the periodic table. The narrative style makes it an excellent companion for students studying chemistry.

3. *Periodic Tales: The Curious Lives of the Elements*

This engaging book explores the stories behind the discovery and use of elements in the periodic table. It combines scientific explanation with historical anecdotes and cultural insights, revealing how chemistry influences everyday life. Readers gain a deeper appreciation for the elements beyond their atomic numbers.

4. *Crash Course Chemistry: Understanding the Periodic Table*

Designed as a supplement to the popular Crash Course video series, this book provides concise explanations and summaries of key concepts related to the periodic table. It is ideal for students seeking a quick yet thorough review of element properties, periodic trends, and chemical behavior. The clear layout and examples make complex ideas easier to grasp.

5. *The Disappearing Spoon: And Other True Tales of Madness, Love, and the History of the World from the*

PERIODIC TABLE OF ELEMENTS

THIS BOOK NARRATES THE INTRIGUING AND SOMETIMES BIZARRE STORIES BEHIND THE ELEMENTS AND THEIR DISCOVERY. PACKED WITH HUMOR AND INSIGHT, IT HIGHLIGHTS THE HUMAN SIDE OF SCIENTIFIC ENDEAVORS CONNECTED TO THE PERIODIC TABLE. PERFECT FOR READERS INTERESTED IN CHEMISTRY AND HISTORY ALIKE.

6. INTRODUCTORY CHEMISTRY: THE PERIODIC TABLE AND BEYOND

A COMPREHENSIVE TEXTBOOK THAT INTRODUCES THE FUNDAMENTALS OF CHEMISTRY WITH A STRONG FOCUS ON THE PERIODIC TABLE'S STRUCTURE AND SIGNIFICANCE. IT EXPLAINS ATOMIC THEORY, ELEMENT CLASSIFICATION, AND PERIODIC TRENDS WITH CLARITY AND DEPTH, SUITABLE FOR BEGINNERS AND HIGH SCHOOL STUDENTS. THE BOOK INCLUDES PRACTICE QUESTIONS AND REAL-WORLD APPLICATIONS TO REINFORCE LEARNING.

7. THE ELEMENTS: A VISUAL EXPLORATION OF EVERY KNOWN ATOM IN THE UNIVERSE

THIS VISUALLY STUNNING BOOK SHOWCASES ALL THE KNOWN ELEMENTS WITH HIGH-QUALITY PHOTOGRAPHS AND INFORMATIVE DESCRIPTIONS. IT HIGHLIGHTS THE NATURAL OCCURRENCE, USES, AND UNIQUE PROPERTIES OF EACH ELEMENT, MAKING THE PERIODIC TABLE COME ALIVE THROUGH IMAGERY. THE BOOK SERVES AS BOTH A REFERENCE AND A COFFEE-TABLE CONVERSATION STARTER.

8. PERIODIC TABLE FOR KIDS: AN EASY GUIDE TO CHEMISTRY

TAILORED FOR YOUNGER AUDIENCES, THIS BOOK SIMPLIFIES THE PERIODIC TABLE AND CHEMICAL CONCEPTS IN AN ENGAGING AND ACCESSIBLE MANNER. COLORFUL ILLUSTRATIONS AND FUN FACTS HELP CHILDREN UNDERSTAND ELEMENTS AND THEIR RELATIONSHIPS WITHOUT OVERWHELMING SCIENTIFIC JARGON. IT ENCOURAGES CURIOSITY AND FOUNDATIONAL KNOWLEDGE IN CHEMISTRY.

9. BEYOND THE PERIODIC TABLE: ADVANCES IN CHEMICAL ELEMENTS AND THEIR IMPACT

THIS BOOK EXPLORES RECENT DISCOVERIES AND DEVELOPMENTS RELATED TO THE PERIODIC TABLE AND ITS ELEMENTS. IT COVERS NEW SYNTHETIC ELEMENTS, CUTTING-EDGE APPLICATIONS, AND THE EVOLVING UNDERSTANDING OF ATOMIC STRUCTURE. IDEAL FOR READERS INTERESTED IN MODERN CHEMISTRY AND THE FUTURE DIRECTION OF ELEMENTAL SCIENCE.

Crash Course Chemistry Periodic Table

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crash course chemistry periodic table: AP Chemistry Crash Course, Book + Online

Adrian Dingle, 2025-04-03 AP Chemistry Crash Course - A Higher Score in Less Time Make the most of your study time and earn a high score with America's bestselling rapid review for AP exams. Here's why more AP students - and teachers - turn to REA's AP Chemistry Crash Course for today's AP digital exam: Targeted Review - Study Only What You Need to Know. Our Crash Course is based on an in-depth analysis of the latest AP Chemistry course and exam description and sample AP digital test questions released by the College Board. It covers only the information tested on the exam, keyed to the nine AP Chemistry units, so you can make the most of your valuable study time. Expert Test-taking Strategies and Advice. Written by Adrian Dingle, an award-winning AP Chemistry teacher and test development expert, the book gives you the topics and critical context that will matter most on exam day. Crash Course relies on the author's extensive analysis of the test's structure and content. By following his advice, you can boost your score. Practice questions - a mini-test in the book, a full-length exam online. Are you ready for your exam? Try our focused practice set inside the book. Then go online to take our full-length practice exam. You'll get the benefits of timed testing, detailed answers, and automatic scoring that pinpoints your performance based on the official AP exam topics - so you'll be confident on test day. Whether you're cramming

for the exam or looking to recap and reinforce your teacher's lessons, Crash Course is the rapid-review study guide every AP student needs. Avoid test anxiety. Go with an author who knows exactly what you need to excel... A true "chemistry freelancer" and subject matter expert, author Adrian Dingle brings more than 30 years of classroom chemistry teaching experience, along with tens of thousands of hours of one-on-one chemistry tutoring around the globe, to a 17-year writing career that includes several bestselling, international award-winning chemistry books and a burgeoning portfolio of other chemistry writing projects.

crash course chemistry periodic table: A Crash Course In Aieee Chemistry 2009 Arora, 2009-09

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crash course chemistry periodic table: CRASH COURSE JEE(MAIN) / AIEEE - CHEMISTRY V&S EDITORIAL BOARD, 2015-01-09 This book is meant to be a quick refresher for JEE (MAIN)/AIEEE aspirants. With the aim and scope of providing a comprehensive study package for aspirants of JEE (MAIN)/AIEEE, this crash course focuses less on theory and more on concepts, formulae and tips. This is supported by plenty of practice problems based on the latest formats, structure and syllabus of JEE (MAIN)/AIEEE. This is further supplemented by a CD given along with this study kit with fully solved 2012 JEE (MAIN)/AIEEE question paper. Salient features: A Based on the latest pattern and syllabus of JEE (MAIN)/AIEEE A Solved examples, practice problems in each chapter A Previous years question papers fully solved A Less theory and more concepts, formulae and tips A Practice CD with fully solved JEE (MAIN)/AIEEE 2012 question paper A Plenty of problems for practice A Comprehensive, holistic revision of the complete syllabus of JEE

(MAIN)/AIEEE A In-depth analysis of the recent trends of JEE (MAIN)/AIEEE A A quick and efficient study kit for JEE (MAIN)/AIEEE aspirants A Facilitates self-study. A Low priced, handy book for quick and efficient revision

crash course chemistry periodic table: AP® Chemistry Crash Course, 2nd Ed., Book + Online Adrian Dingle, 2014-02-13 REA's Crash Course for the AP® Chemistry Exam - Gets You a Higher Advanced Placement® Score in Less Time Crash Course is perfect for the time-crunched student, the last-minute studier, or anyone who wants a refresher on the subject. Are you crunched for time? Have you started studying for your Advanced Placement® Chemistry exam yet? How will you memorize everything you need to know before the test? Do you wish there was a fast and easy way to study for the exam AND boost your score? If this sounds like you, don't panic. REA's Crash Course for AP® Chemistry is just what you need. Our Crash Course gives you: Targeted, Focused Review - Study Only What You Need to Know Fully revised for the 2014 AP® Chemistry exam, this Crash Course is based on an in-depth analysis of the revised AP® Chemistry course description outline and sample AP® test questions. It covers only the information tested on the new exam, so you can make the most of your valuable study time. Our targeted review focuses on the Big Ideas that will be covered on the exam. Explanations of the AP® Chemistry Labs are also included. Expert Test-taking Strategies This Crash Course presents detailed, question-level strategies for answering both the multiple-choice and essay questions. By following this advice, you can boost your score in every section of the test. Take REA's Online Practice Exam After studying the material in the Crash Course, go to the online REA Study Center and test what you've learned. Our practice exam features timed testing, detailed explanations of answers, and automatic scoring analysis. The exam is balanced to include every topic and type of question found on the actual AP® exam, so you know you're studying the smart way. Whether you're cramming for the test at the last minute, looking for extra review, or want to study on your own in preparation for the exams - this is the study guide every AP® Chemistry student must have. When it's crucial crunch time and your Advanced Placement® exam is just around the corner, you need REA's Crash Course for AP® Chemistry! About the Author Adrian Dingle is a chemistry educator and author, with 24 years of experience teaching in the United States and the United Kingdom. He is the creator of the award-winning chemistry website, www.adriandingleschemistrypages.com. The focus of Mr. Dingle's teaching career has been on preparing students for standardized tests; AP® and SAT® tests in the United States, GCSE's and A levels in the United Kingdom, and International Baccalaureate in both countries. An Englishman, he lives in Atlanta, Georgia, where he teaches at The Westminster Schools. He holds a B.Sc. (Hons.) Chemistry, and a Postgraduate Certificate in Education, both from the University of Exeter in England. In addition to writing this Crash Course, Mr. Dingle has written The Periodic Table: Elements With Style, How To Make A Universe With 92 Ingredients, and SAT™ Chemistry Crash Course. He is the 2011 winner of the School Library Association of the UK's Information Book Award, and, in 2012, was honored with the prestigious literary prize Wissenschaftsbuch des Jahre, sponsored by the Austrian Ministry of Science and Research.

crash course chemistry periodic table: AP® Chemistry Crash Course, For the 2020 Exam, Book + Online Adrian Dingle, 2020-02-04 This Crash Course book is designed to be an aide for someone working to pass the AP (advanced placement) test in Chemistry--

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crash course chemistry periodic table: SAT Subject Test: Chemistry Crash Course Adrian Dingle, 2013-07-02 SAT* Chemistry Subject Test Crash Course - Gets You a Higher Score in Less Time Our Crash Course is perfect for the time-crunched student, the last-minute studier, or anyone who wants a refresher on the subject. Are you crunched for time? Have you started studying for your SAT* Chemistry Subject Test yet? How will you memorize everything you need to know before the exam? Do you wish there was a fast and easy way to study for the test AND raise your score? If this sounds like you, don't panic. SAT* Chemistry Crash Course is just what you need. Crash Course gives you: Targeted, Focused Review - Study Only What You Need to Know The Crash Course is based on an in-depth analysis of the SAT* Chemistry course description and actual test questions. It

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crash course chemistry periodic table: 7 Days JEE Main Crash Course for Organic Chemistry I Disha Experts, 2020-02-04

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